

SHORT COMMUNICATION

AN ATTEMPT TO FIND EVIDENCE FOR THE EXISTENCE OF (+)-GIBBERELLIC ACID

B. E. CROSS, K. NORTON and J. C. STEWART

Department of Organic Chemistry, The University, Leeds 2

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Abstract—[17-¹⁴C](+)-Kaurene has been prepared and added to a fermentation of *Gibberella fujikuroi*; it was not incorporated into gibberellic acid.

INTRODUCTION

THE gibberellins are a class of endogenous plant hormones which regulate various aspects of plant growth and development.^{1,2} Their occurrence is so widespread that it is possible that they are to be found in all higher plants.¹ Many of the gibberellins are also produced by the fungus *Gibberella fujikuroi*³ and it has been shown⁴ that (–)-gibberellic acid (I), the most readily available gibberellin, can be biosynthesized by the fungus from the diterpene (–)-kaurene (III) which is known to occur in several trees.⁵ Furthermore with *G. fujikuroi*, the closely related (–)-kauren-19-oic acid (IV) has also been shown⁶ to act as a precursor of gibberellins A₁ (VI), A₄ (VII) and A₇ (II), whilst (–)-kauren-19-ol (V) is transformed⁷ into gibberellin A₁₃ (VIII). It is generally believed that the biosynthesis of the gibberellins in plants follows a similar pathway.^{3,8,9}

All the known gibberellins are believed to belong to the same stereochemical series, i.e. to the diterpenes with an antipodal A/B ring fusion. However the widespread occurrence of diterpenes of the enantiomeric series ('normal' A/B ring fusion) in higher plants and in particular the isolation of (+)-kaurene from *Podocarpus spicatus*¹⁰ and *P. ferrugineus*¹¹ suggests that some plants may contain gibberellins which are the enantiomers of those so far reported. Since the isolation of gibberellins from a tree such as *P. spicatus* would almost certainly be long and tedious and the amount of gibberellin obtained might be inadequate for complete characterization we tried another approach to the search for (+)-gibberellins. The possibility that the enzyme systems of *G. fujikuroi* would utilize (+)-kaurene as a precursor of

¹ P. W. BRIAN, *Intern. Rev. Cytol.* **19**, 229 (1966).

² L. G. PALEG, *Ann. Rev. Plant Physiol.* **16**, 291 (1965).

³ B. E. CROSS, *Progr. Phytochem.* in press.

⁴ B. E. CROSS, R. H. B. GALT and J. R. HANSON, *J. Chem. Soc.* 295 (1964).

⁵ R. T. APLIN, R. C. CAMBIE and P. S. RUTLEDGE, *Phytochem.* **2**, 205 (1963) and references cited therein.

⁶ T. A. GEISSMAN, A. J. VERBISCAR, B. O. PHINNEY and G. CRAGG, *Phytochem.* **5**, 933 (1966).

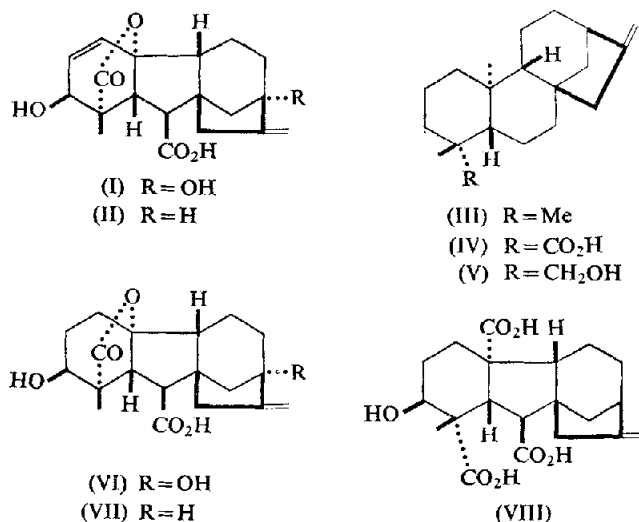
⁷ B. E. CROSS, K. NORTON and R. H. B. GALT, *Tetrahedron* **23**, in press.

⁸ J. E. GRAEBE, D. T. DENNIS, C. D. UPPER and C. A. WEST, *J. Biol. Chem.* **240**, 1847 (1965).

⁹ D. T. DENNIS and C. A. WEST, *J. Biol. Chem.* **242**, 3293 (1967).

¹⁰ J. R. MCGIMPSEY and J. MURRAY, *J. Appl. Chem.* **10**, 340 (1960).

¹¹ L. H. BRIGGS, R. W. CAWLEY, J. A. LOE and W. I. TAYLOR, *J. Chem. Soc.* 955 (1950).



(+)-gibberellic acid was clearly very remote, however the consequences of success would be very important and [17-¹⁴C](+)-kaurene is readily prepared. Labelled (+)-kaurene was therefore added to a fermentation of the fungus in the usual way.^{4,7} The gibberellic acid produced was isolated and purified as its methyl ester but it was not radioactive. Consequently in order to determine whether gibberellins of the (+)-series exist it would appear to be necessary to isolate the gibberellins from a (+)-kaurene producing plant.

EXPERIMENTAL

(General experimental methods have been previously described.⁷)

[17-¹⁴C](+)-Kaurene

(+)-Kaurene (m.p. 47–48°, [α]_D+80° (c 0.56); 150 mg) was converted into [17-¹⁴C](+)-kaurene (48 mg) by the method used⁴ to prepare [17-¹⁴C](–)-kaurene.

Addition of [17-¹⁴C](+)-kaurene to a *G. fujikuroi* fermentation

G. fujikuroi ACC 917 was grown on a Cerelease-ammonium tartrate medium until the ammonium nitrogen was exhausted. [17-¹⁴C](+)-Kaurene (18 mg; 4.06 μ c) in EtOH (20 ml) was added and the fermentation continued for 112 hr. The fermentation was worked-up as previously described⁷ to give crude gibberellic acid (540 mg) which, after methylation with ethereal diazomethane and purification,⁷ gave methyl gibberellate (296 mg; 58 counts/100 sec/m-mole) which was identified by its i.r. spectrum. TLC of the crude gibberellic acid on silica gel G in di-isopropyl ether-AcOH (95:5) followed by autoradiography⁴ failed to reveal any radioactive spots.

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